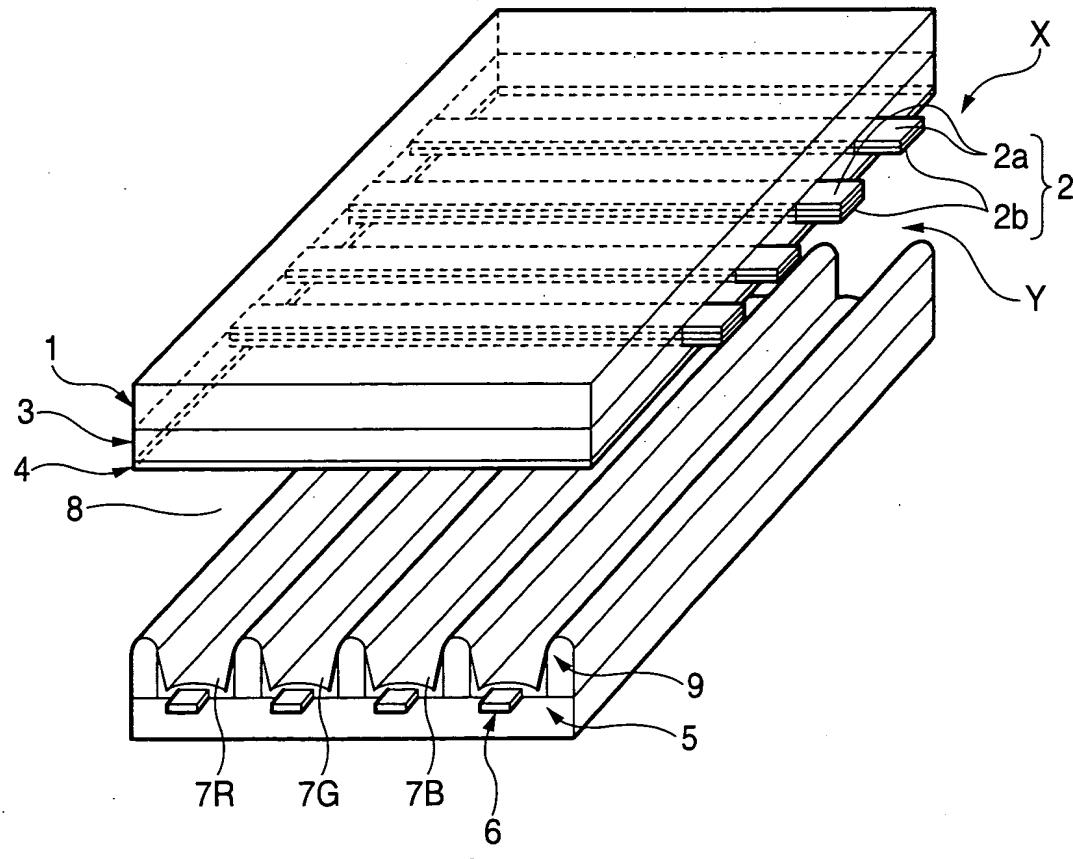


FIG. 1
PRIOR ART



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FIG. 2 PRIOR ART

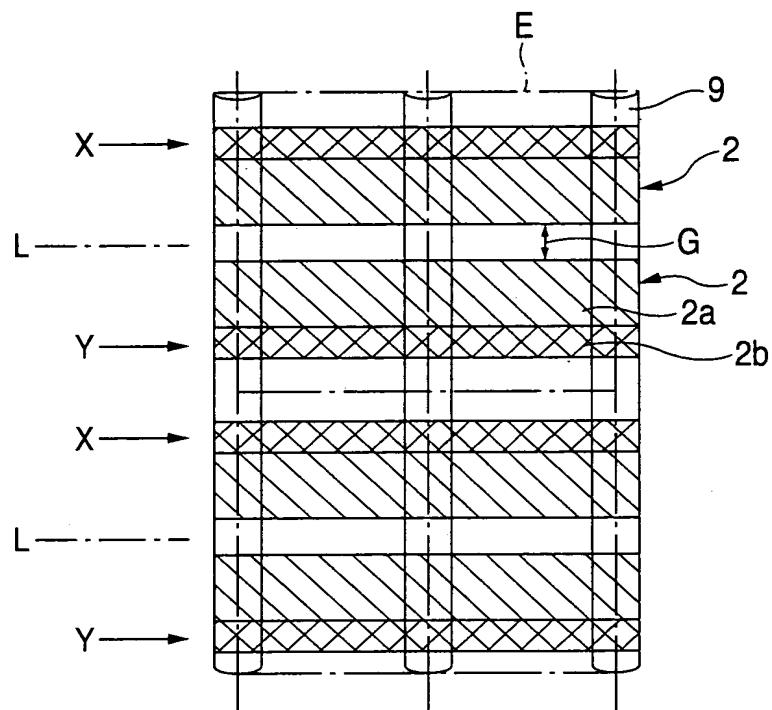


FIG. 3 PRIOR ART

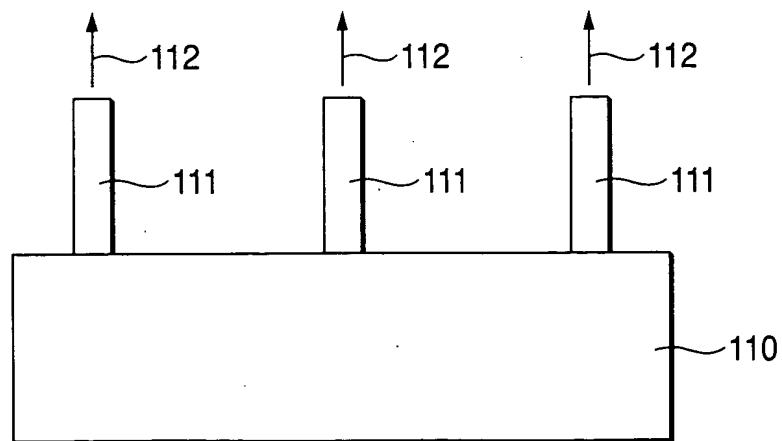


FIG. 4

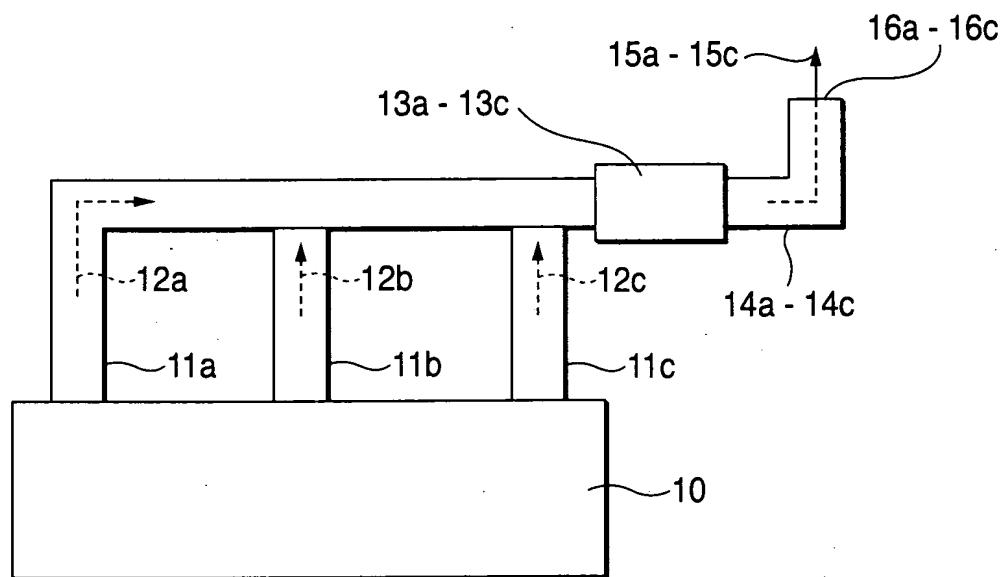


FIG. 5

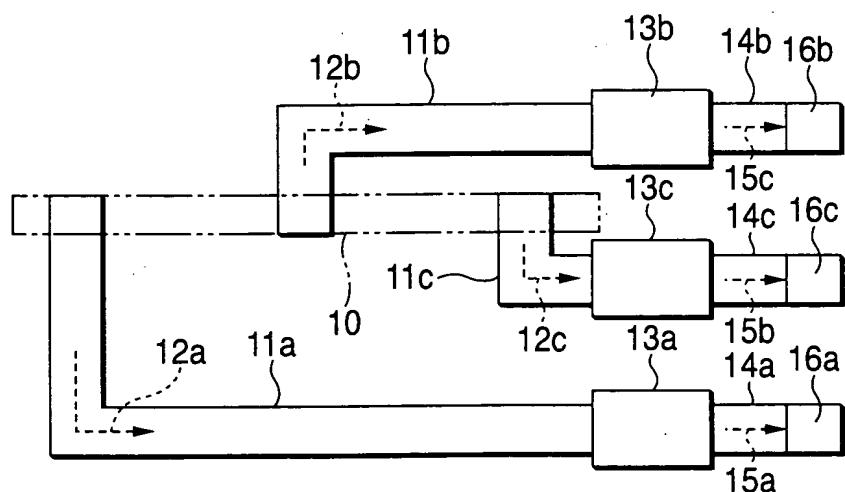


FIG. 6

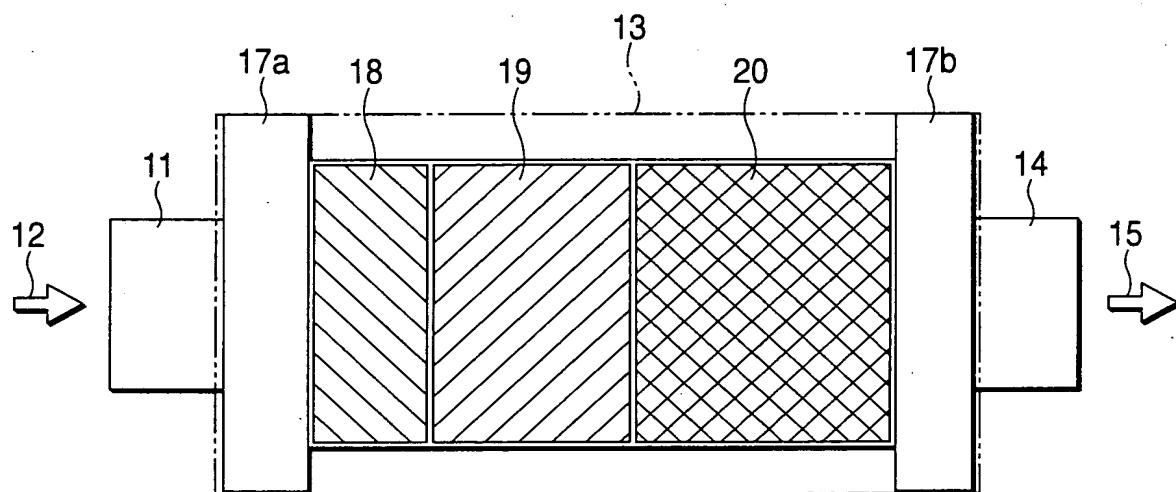


FIG. 7

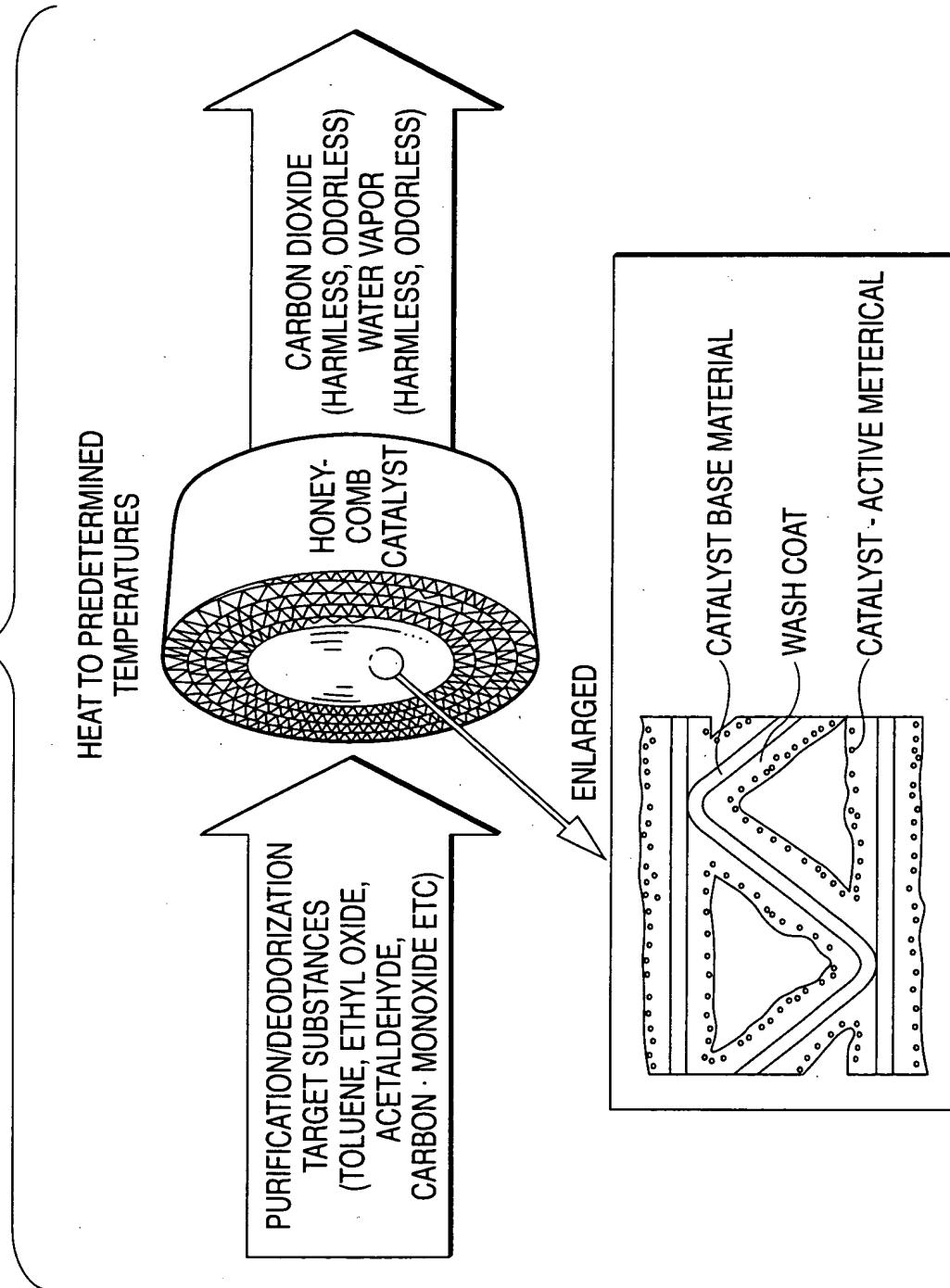
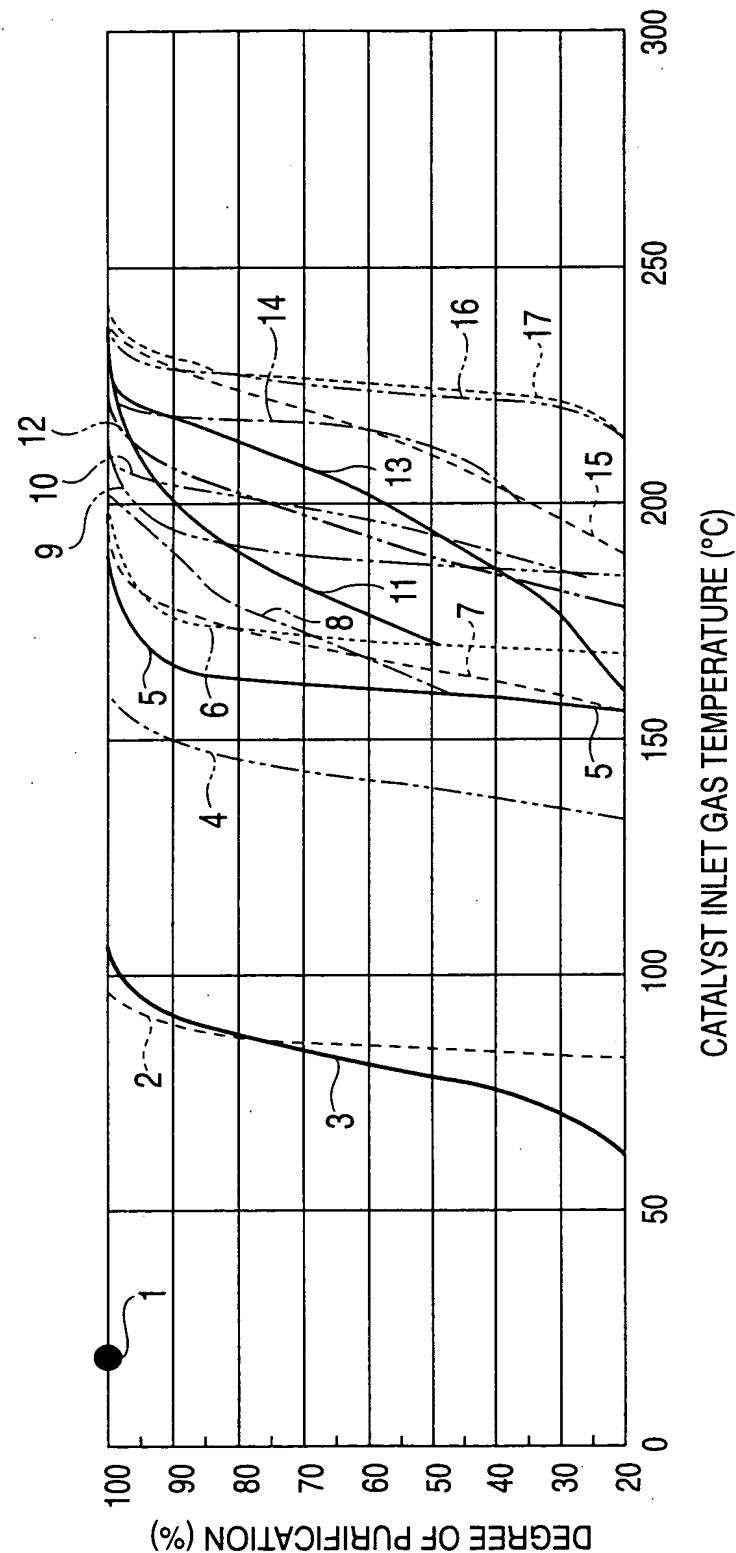


FIG. 8 (a)

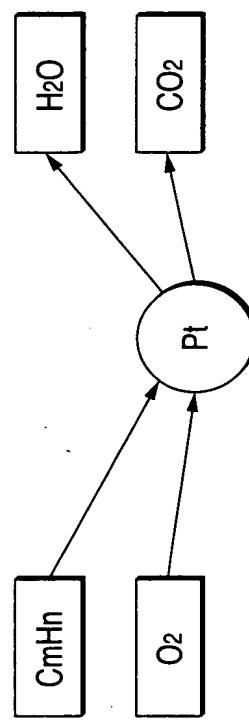


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FIG. 8 (b)

NO.	SUBSTANCE NAME	CHEMICAL FORMULA	CONCENTRATION (ppm)	SPACE VELOCITY (h ⁻¹)
1		H ₂	1%	60,000
2		CO	1,000	60,000
3		CH ₃ OH	100	30,000
4		C ₂ H ₄	5,000	60,000
5		C ₆ H ₁₀ O	550	60,000
6		C ₆ H ₅ CH ₃	550	60,000
7		C ₂ H ₅ COCH ₃	650	60,000
8		(CH ₃) ₂ S	10	30,000
9		C ₆ H ₄ (CH ₃) ₂	550	60,000
10		NH ₃	300	30,000
11		(CH ₃) ₃ N	30	30,000
12		CH ₃ CHO	140	30,000
13		C ₂ H ₅ OH	300	30,000
14		CH ₃ C ₆ H ₄ OH + C ₆ H ₅ OH	660 + 440	60,000
15		(C ₂ H ₅) ₃ N	300	30,000
16		CH ₃ COOH	100	30,000
17		HCON(CH ₃) ₂	740	60,000

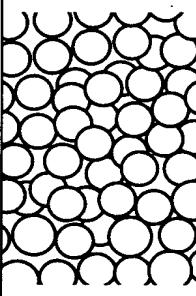
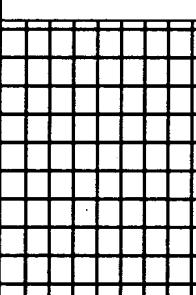
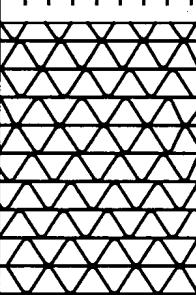
REACTION FORMULA: $C_mH_n + O_2 \rightarrow CO_2 + H_2O + \text{REACTION HEAT (EXOTHERMAL REACTION)}$



TEMPERATURE RISE RESULTING FROM REACTION HEAT
(1000 PPM CONCENTRATION GAS COMBUSTION) = $10 + 20 \times C (\text{CARBON}) \text{ NUMBER (}^{\circ}\text{C)}$

FIG. 9

FIG. 10

CATALYST TYPE	METAL HONEYCOMB CATALYST	CERAMIC HONEYCOMB CATALYST	PELLET CATALYST
CATALYST TYPE			
BASIC COMPOSITION	Fe-Cr-Al	SiO ₂ -Al ₂ O ₃ -MgO	Y-Al ₂ O ₃
COEFFICIENT OF HEAT CONDUCTIVITY	LARGE	SMALL	SMALL
FILLED SPECIFIC GRAVITY	0.4 TO 0.6	0.6 TO 0.7	0.4 TO 0.8
HEAT CAPACITY	SMALL	MODERATE	LARGE
STANDARD SV VALUE	30,000 TO 60,000 h ⁻¹	20,000 TO 40,000 h ⁻¹	10,000 TO 30,000 h ⁻¹
PRESSURE LOSS (*)	5.5	7.1	41.5
MECHANICAL STRENGTH	STRONG	WEAK	MODERATE
THERMAL SHOCK RESISTANCE	STRONG	WEAK	MODERATE

(*: MEASURED VALUE UNDER AN ATMOSPHERE OF 200°C AND 1 Nm/s.)